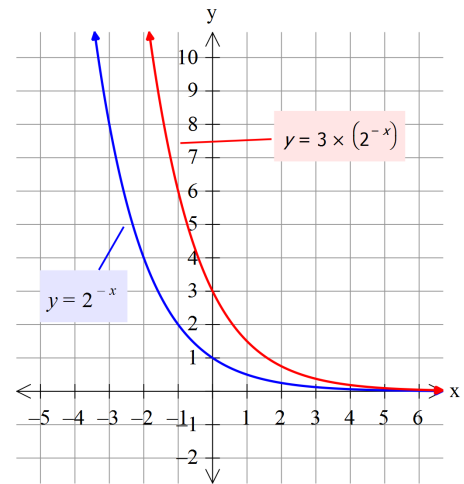
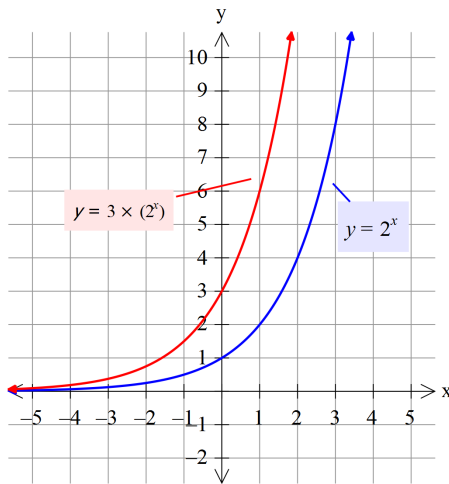


Exponential Functions

Here the independent variable (x) is the exponent of the function

$$y = ka^x$$

Examples



- The y -intercept is k
- a determines the steepness
- $y = ka^{-x}$ is a reflection of $y = ka^x$ in the y -axis
- In all these graphs the x -axis is an asymptote.

If $a > 1$ (ie base a is positive) the equation $y = ka^x$ models exponential growth

If $a < 1$ (ie base a is negative) the equation $y = ka^x$ models exponential decay

(N.B. this is the same as having $a > 1$ and graphing $y = ka^{-x}$)

Applications: compound interest

\$2000 is invested at 4% p.a. for 10 years. What is the current value (v) after the 10 years?

Let the initial amount invested (principle) be P . Look at the current value (v)

After 1 year

$$\begin{aligned} v &= P + 0.04P \\ &= P(1 + 0.04) \\ &= P1.04 \end{aligned}$$

After 10 years

$$\begin{aligned} v &= 2000(1.04)^{10} \\ &= 2960.49 \end{aligned}$$

After 2 years

$$\begin{aligned} v &= P1.04 \times 1.04 \\ &= P(1.04)^2 \end{aligned}$$

After 3 years

$$\begin{aligned} v &= P(1.04)^2 \times 1.04 \\ &= P(1.04)^3 \end{aligned}$$

After n years

$$v = P(1.04)^n$$



Applications: radioactive decay

The decay of Carbon-14 can be modelled using the equation

$$N(t) = N_0 e^{-0.000121t} \quad \text{where } N = \text{the current amount, } t = \text{time (years)}$$

and $N_0 =$ the original amount

(Equation for $N(t)$ is from the web site <http://physics.bu.edu/~duffy/py106/Radioactivity.html>)

- (a) If you start with 2 kg of C-14, how much will you have in 50 years, 1000 years and 3000 years?
- (b) How long will it take you to get down to 1 kg?

Solution

(a) Mass of C-14 after ...

In 50 years

$$\begin{aligned} N &= 2e^{-0.000121 \times 50} \\ &= 2e^{-0.00605} \\ &= 1.988 \end{aligned}$$

In 1000 years

$$\begin{aligned} N &= 2e^{-0.000121 \times 1000} \\ &= 2e^{-0.121} \\ &= 1.772 \end{aligned}$$

In 3000 years

$$\begin{aligned} N &= 2e^{-0.000121 \times 3000} \\ &= 2e^{-0.363} \\ &= 1.391 \end{aligned}$$

(b)

How long will it take you to get down to 1 kg?

We need

$$\begin{aligned} 1 &= 2e^{-0.000121t} \\ 0.5 &= e^{-0.000121t} \\ \ln 0.5 &= -0.000121t \\ \frac{\ln 0.5}{-0.000121} &= t \\ t &= 5728 \text{ years} \end{aligned}$$

Exercises

- \$5000 is invested at an interest rate of 5% p.a. What is the value of the investment
 - after 4 years
 - after 10 years
- \$1500 is invested at an interest rate of 8% p.a.
 - How much will the investment be worth after 5 years?
 - How long will it take for the investment to be worth \$5000?
- The decay of Fluorine 18 can be modelled with the equation

$$N = N_0 e^{-0.381t}$$
 where t is measured in hours

If you had 5 grams of Fluorine 18 :

 - how much would be left after 3 hours?
 - how much would be left after 6 hours?
 - How long would it take you to have only 1 gram left?



(Half-life of Flourine 18 from <http://science.howstuffworks.com/nuclear2.htm>)

4. Radioactive decay of Americium 241 is can be modelled by

$$N = N_0 e^{-0.0015 t} \quad \text{where } t \text{ is measured in years}$$

If you had 1 kg of Americium 241:

- how much would be left after 10 years?
- how much would be left after 20 years?
- How long would it take you to have only 500 grams left?

(Americium 241 half-life from <http://science.howstuffworks.com/nuclear2.htm>)

Answers

1. (a) \$6077.53

(b) \$8144.47

2. (a) \$2203.99

(b)

$$5000 = 1500(1.08)^n$$

$$\frac{5000}{1500} = 1.08^n$$

$$\ln\left(\frac{10}{3}\right) = n \ln 1.08$$

$$\frac{\ln\left(\frac{10}{3}\right)}{\ln 1.08} = n$$

$$n \approx 15.6 \text{ years}$$

3. (a) 1.59g

(b) 0.51g

3. (c)

$$1 = 5e^{-0.381 \times t}$$

$$\frac{1}{5} = e^{-0.381 \times t}$$

$$\ln 0.2 = -0.381 \times t$$

$$t = \frac{\ln 0.2}{-0.381}$$

$$t \approx 4.22 \text{ hours}$$

4. (a) 985g

(b) 970g

(c)

$$500 = 1000e^{-0.0015 \times t}$$

$$\frac{500}{1000} = e^{-0.0015 \times t}$$

$$\ln 0.5 = -0.0015 \times t$$

$$t = \frac{\ln 0.5}{-0.0015}$$

$$t = 462 \text{ years}$$