



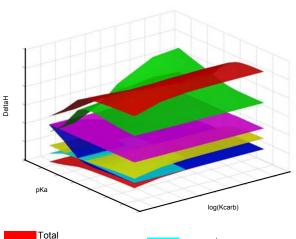
RESEARCH LABORATORY

рΗ

Research Focus

Amongst several other, non-energy related projects, the Chemometrics laboratory is currently investigating the fundamental understanding of the interaction of CO2 with organic amines and other bases in aqueous solution. This research is relevant for postcombustion capture (PCC) of CO₂ in coal fired power stations and several other applications. There is a strong collaboration with CSIRO Energy Technology in Newcastle and the Swiss Federal Institute of Technology (ETH).

RNHCO RNHCO₂H +RNH₂ $CO_2(aq)$ H₂CO₃ HCO



A + H⁺↔AH⁺

 $CO_2(g) \leftrightarrow CO_2(aq)$

A+HCO₃² ↔ACO₂

$H_2O \leftrightarrow H^+ + OH^-$ HCO₃²⁻ ↔ H⁺ + CO₃²⁻ CO₂(aq)↔H⁺ + HCO₃

Current Projects

The immediate goal of our research is to determine all relevant rates and equilibrium constants for all reactions between CO₂ and organic amines. A complete reaction scheme for monobasic amines is given in the scheme to the right. The results will include the temperature dependences of all constants as well as the heat capacities of all reactions. The second goal is to model the performance of single amine and amine mixture solutions in pilot plant situations. The modelling results will continuously be compared with laboratory and pilot plant absorbers at CSIRO. Plant modelling is based on molecular mechanisms and not on empirical reactions which are traditionally used in the engineering contexts. It is our belief that this more scientific approach will advance the understanding of CO₂ amine interactions in significant ways; e.g. it allows the detailed analysis of all components of the overall interaction of CO₂ with amines as a function of amine parameters such as pK_a-values, reaction enthalpies, carbamate formation, etc. The graph displays the resolution of the overall cyclic reaction enthalpy into the different components as a function of amine protonation and carbamate formation constant. PCC is done in concentrated solutions, thus the incorporation of activity coefficients is crucial. In a related project we are investigating different activity coefficient approximations for their applicability in situations relevant for PCC. Software for all the above computations is developed in Matlab.

Equipment

The chemometrics laboratory has a complete range of instrumentation for the investigation of chemical processes in solution; this includes in particular kinetic and equilibrium studies. Spectroscopic instrumentation includes NMR facilities, UV-Vis and IR spectrometers as well as access to MS equipment; analytical instrumentation: HPLC, GC, AA and ICP; general lab facilities with fume cupboards for synthetic work. In addition to this, the laboratory has several fully computerised titration set-ups that include potentiometric and spectrophotometric instruments for equilibrium studies and a stopped-flow instrument with diode-array detection for fast kinetic studies. Currently, the laboratory, in collaboration with Dr Graeme Puxty of CSIRO, is in the process of building a research calorimeter designed by Professor Hungerbühler at the ETH. Most relevant is a complete range of software for the modelling and analysis of chemical processes in solution.

Contact Us

PRIORITY RESEARCH CENTRE FOR ENERGY

FOR INFORMATION ABOUT THE CHEMOMETRICS RESEARCH LABORATORY CONTACT Associate Professor Marcel Maeder
Faculty of Science and Information Technology
School of Environmental and Life Sciences
T +61 2 4921 5478
F +61 2 4921 5472
E Marcel.Maeder@newcastle.edu.au

Dr Nichola McCann
Faculty of Science and Information Technology
School of Environmental and Life Sciences
T +61 2 4921 8737
F +61 2 4921 5472
E Nichola.Mccann@newcastle.edu.au

FOR GENERAL INFORMATION ABOUT THE CENTRE CONTACT Mr Shannon Martin
Centre Coordinator
Faculty of Engineering and Built Environment
The University of Newcastle
Callaghan, NSW, Australia 2308
T + 61 2 4921 5466
F + 61 2 4921 6893
E Shannon.Martin@newcastle.edu.au

livesite.newcastle.edu.au/energy

